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[NAMING IONIC COMPOUNDS WORKSHEET 2](#)

Naming Ionic Compounds Worksheet 2 W 308 Name the following ionic compounds: 1) $\text{Mo}(\text{SO}_4)_3$ 2) $\text{Ti}(\text{NO}_3)_3$ 3) NaH 4) $\text{BeSO}_3 \cdot 7\text{H}_2\text{O}$ 5) $\text{Fe}_3(\text{PO}_4)_2$ 6) $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ 7) $\text{V}(\text{H}_2\text{PO}_4)_4$ 8) $\text{Mn}_3(\text{PO}_4)_7$ 9) $\text{Fe}(\text{CN})_3$ 10) BaSO_4 Write the formulas for the following ionic compounds: 11) vanadium (V) arsenide 12) cobalt (III) nitrite 13) yttrium perchlorate 14) silver phosphide 15) tin (IV) selenide 16) cadmium sulfite 17) nickel (III)...

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Lab 2: Naming Ionic Compounds Vocabulary cations positively charged ions anions negatively charged ions cations anions Na^+ Ca^{2+} NH_4^+ Cl^- NO_3^- OH^- Ag^+ Al^{3+} Mg^{2+} SO_4^{2-} PO_4^{3-} ...

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