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a. What mass of water is needed to react with 32.9 g Li_3N ? (51.06 g H_2O)
b. When 32.9 g Li_3N are used, how many molecules of NH_3 are produced? (5.69 $\times 10^{23}$ molecules NH_3) ...

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1. If you earned one penny every 10 seconds of your life, how many dollars would you have after 65 years? (\$2,049,840)
2. Calculate the number of moles in the following.
a. 7.85 g of Fe (0.141 mol)
b. 6.55×10^{13} atoms of ^{14}C (1.09 $\times 10^{10}$ mol)
3. In 1992, the United States manufactured 2.47×10^8 pounds of phosphoric acid (H_3PO_4). How m...

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